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Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

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Office Action Summary	Application No. 10/593,960	Applicant(s) ARAKI, YASUSHI	
	Examiner RAJ GUPTA	Art Unit 2814	

-- The MAILING DATE of this communication appears on the cover sheet with the correspondence address --

Period for Reply

A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE 3 MONTH(S) OR THIRTY (30) DAYS, WHICHEVER IS LONGER, FROM THE MAILING DATE OF THIS COMMUNICATION.

- Extensions of time may be available under the provisions of 37 CFR 1.136(a). In no event, however, may a reply be timely filed after SIX (6) MONTHS from the mailing date of this communication.
- If NO period for reply is specified above, the maximum statutory period will apply and will expire SIX (6) MONTHS from the mailing date of this communication.
- Failure to reply within the set or extended period for reply will, by statute, cause the application to become ABANDONED (35 U.S.C. § 133). Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patent term adjustment. See 37 CFR 1.704(b).

Status

- 1) ☒ Responsive to communication(s) filed on 28 July 2010.
- 2a) ☒ This action is **FINAL**. 2b) ☐ This action is non-final.
- 3) ☐ Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under *Ex parte Quayle*, 1935 C.D. 11, 453 O.G. 213.

Disposition of Claims

- 4) ☒ Claim(s) 1,4,12 and 14-20 is/are pending in the application.
- 4a) Of the above claim(s) _____ is/are withdrawn from consideration.
- 5) ☐ Claim(s) _____ is/are allowed.
- 6) ☒ Claim(s) 1,4,12 and 14-20 is/are rejected.
- 7) ☐ Claim(s) _____ is/are objected to.
- 8) ☐ Claim(s) _____ are subject to restriction and/or election requirement.

Application Papers

- 9) ☐ The specification is objected to by the Examiner.
- 10) ☐ The drawing(s) filed on _____ is/are: a) ☐ accepted or b) ☐ objected to by the Examiner.
Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).
Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d).
- 11) ☐ The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152.

Priority under 35 U.S.C. § 119

- 12) ☐ Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).
- a) ☐ All b) ☐ Some * c) ☐ None of:
1. ☐ Certified copies of the priority documents have been received.
2. ☐ Certified copies of the priority documents have been received in Application No. _____.
3. ☐ Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).

* See the attached detailed Office action for a list of the certified copies not received.

Attachment(s)

- | | |
|--|---|
| 1) <input checked="" type="checkbox"/> Notice of References Cited (PTO-892) | 4) <input type="checkbox"/> Interview Summary (PTO-413)
Paper No(s)/Mail Date. _____ |
| 2) <input type="checkbox"/> Notice of Draftperson's Patent Drawing Review (PTO-948) | 5) <input type="checkbox"/> Notice of Informal Patent Application |
| 3) <input type="checkbox"/> Information Disclosure Statement(s) (PTO/SB/08)
Paper No(s)/Mail Date _____ | 6) <input type="checkbox"/> Other: _____ |

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Attorney's Docket Number: Q97019

Filing Date: 3/18/2005

371 Date: 9/22/2006

Claimed Domestic Priority: NONE

Claimed Foreign Priority: 3/22/2004 (JP 2004-082002)

Applicant: Araki

Examiner: Raj R. Gupta

DETAILED ACTION

This Office Action responds to the amendment filed on 7/28/2010.

Acknowledgement

1. The amendment filed on 7/28/2010, responding to the Office Action mailed on 4/28/2010, has been entered. The present Office Action is made with all the suggested amendments being fully considered. Accordingly, pending in this Office Action are **claims 1, 4, 12, and 14-20**.

Claim Rejections - 35 USC § 103

2. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negated by the manner in which the invention was made.

3. Initially, and with respect to **claims 15 and 16**, note that a “product-by-process” claim is directed to the product *per se*, no matter how actually made. See In re Thorpe et al., 227 USPQ 964 (CAFC, 1985) and the related case law cited therein with make it clear that it is the final

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product *per se* which must be determined in a “product-by-process” claim, and the patentability of the process, and that, as here, an old or obvious product produced by a new method is patentable as a product, whether claimed in “product-by-process” claims or not. As stated in Thorpe,

even though product-by-process claims are limited by and defined by the process, determination of patentability is based on the product itself. *In re Brown*, 459 F.2d 531, 535, 173 USPQ 685, 688 (CCPA 1972); *In re Pilkington*, 411 F.2d 1345, 1348, 162 USPQ 145, 147 (CCPA 1969); *Buono v. Yankee Maid Dress Corp.*, 77 F.2d 274, 279, 26 USPQ 57, 61 (2d. Cir. 1935).

Note that the Applicant has the burden of proof in such cases as the above case law makes clear.

4. **Claims 1, 4, 12, and 14-17** are rejected under 35 U.S.C. 103(a) as being unpatentable over **Petritsch et al (US 6340789)** in view of **Stossel et al (US 7223484)** as evidenced by **Asfandiarov et al (Investigation of Electron Structure of 2,1,3-Benzothiadiazole Derivatives by Means of Negative Ion Mass Spectrometry, Photoelectron Spectroscopy and Absorption Spectroscopy; Rapid Commun. Mass Spectrom. 12, 595–602, 1998)**, **Nakaya et al (US 5792557)**, and **Ise et al. (US 2002/0028329)**.

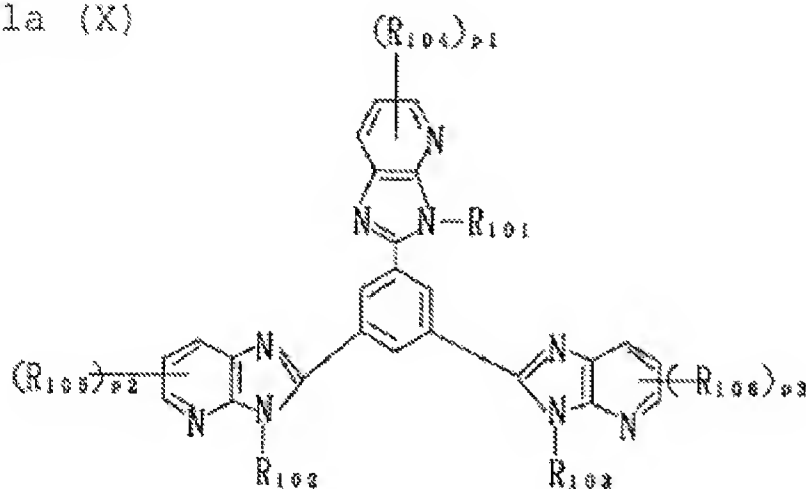
5. With regard to **claim 1**, Petritsch teaches, in Fig 5, a photodetector (Fig 5 item 20) comprising: at least one electron transporting organic material (Fig 5 item 14; “MCP” col 6 ln 31); and at least one hole transporting material (Fig 5 item 6).

6. Petritsch does not explicitly teach that said at least one electron transporting organic material has an ionization potential of 5.8 eV or more, wherein said ionization potential of said at least one electron transporting organic material is larger than an energy necessary for the highest-level electron of said at least one hole transporting organic material to be taken out to a vacuum

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infinite far point, wherein said ionization potential of said at least one electron transporting organic material is larger than an ionization potential of said at least one hole transporting organic material by 0.6 eV or more, and wherein said at least one electron transporting organic material is a compound represented by formula (X):

Formula (X)

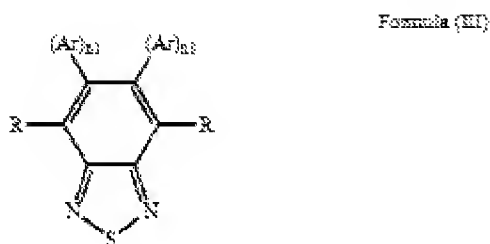


wherein R101, R102 and R103 each represent a hydrogen atom, an aliphatic hydrocarbon group, an aryl group or a hetero ring group, wherein R104, R105, and R106 each independently represents a substituent, wherein the substituent is an alkyl group, an alkenyl group, an alkynyl group, an aryl group, an amino group, an alkoxy group, an aryloxy group, an acyl group, an alkoxycarbonyl group, an aryloxycarbonyl group, an acyloxy group, an acylamino group, an alkoxycarbonylamino group, an aryloxycarbonylamino group, a sulfonylamino group, a sulfamoyl group, a carbamoyl group, an alkylthio group, an arylthio group, a sulfonyl group, a sulfinyl group, a ureido group, a phosphoric acid amido group, a hydroxyl group, a mercapto group, a halogen atom, a cyano group, a sulfo group, a carboxyl group, a nitro group, a

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hydroxamic acid group, a sulfinio group, a hydrazine group, an imino group, a hetero ring group, or a silyl group, wherein p1, p2 and p3 each independently represents an integer of 0 to 3.

7. Stossel teaches the use of 2,1,3-Benzothiadiazole derivatives as Formula (III):



as, "An organic photodetector ... transport material ..." (col 30 ln 11-13), in since, " The 2,1,3-benzothiadiazole-containing compounds of the invention lead, when appropriate devices are used, to excellent operating lives ..." (col 2 ln 57-59).

8. However Stossel does not explicitly disclose the fact that the ionization potential of 2,1,3-Benzothiadiazole derivatives is greater than 5.8 eV.

9. Asfandiarov provides evidence that the ionization energies of 2,1,3-benzothiadiazole derivatives are in excess of 5.8 eV, as they range from 7.77 - 8.44 eV (pg 597, Table 2).

10. Thus Stossel does teach: the ionization potential of said at least one electron transporting organic material is more than 5.8 eV.

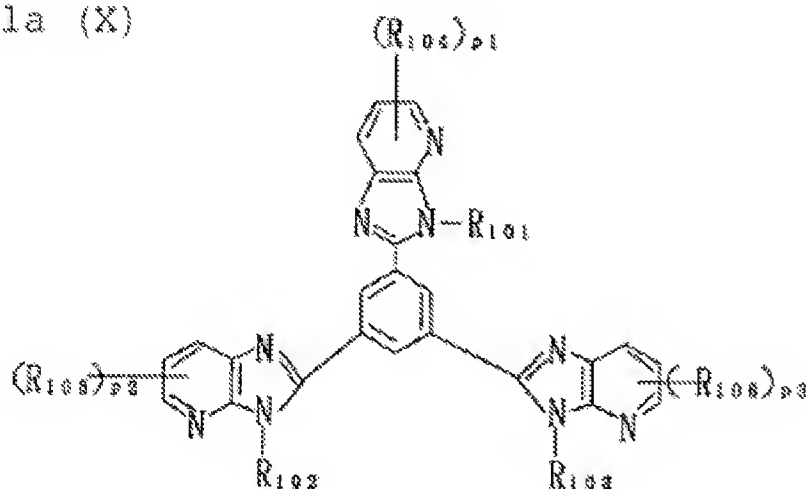
11. Therefore it would have been obvious to one of ordinary skill in the art at the time of the invention to use the electron transporting organic material of Stossel as evidenced by Asfandiarov in concert with the photodetector of Petritsch to lead to appropriate devices with excellent operating lives.

12. Petritsch/Stossel do not explicitly teach that said ionization potential of said at least one electron transporting organic material is larger than an energy necessary for the highest-level

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electron of said at least one hole transporting organic material to be taken out to a vacuum infinite far point, wherein said ionization potential of said at least one electron transporting organic material is larger than an ionization potential of said at least one hole transporting organic material by 0.6 eV or more, and wherein said at least one electron transporting organic material is a compound represented by formula (X):

Formula (X)



wherein R_{101} , R_{102} and R_{103} each represent a hydrogen atom, an aliphatic hydrocarbon group, an aryl group or a hetero ring group, wherein R_{104} , R_{105} , and R_{106} each independently represents a substituent, wherein the substituent is an alkyl group, an alkenyl group, an alkynyl group, an aryl group, an amino group, an alkoxy group, an aryloxy group, an acyl group, an alkoxycarbonyl group, an aryloxycarbonyl group, an acyloxy group, an acylamino group, an alkoxycarbonylamino group, an aryloxycarbonylamino group, a sulfonylamino group, a sulfamoyl group, a carbamoyl group, an alkylthio group, an arylthio group, a sulfonyl group, a sulfinyl group, a ureido group, a phosphoric acid amido group, a hydroxyl group, a mercapto group, a halogen atom, a cyano group, a sulfo group, a carboxyl group, a nitro group, a

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hydroxamic acid group, a sulfinio group, a hydrazine group, an imino group, a hetero ring group, or a silyl group, wherein p1, p2 and p3 each independently represents an integer of 0 to 3.

13. Nakaya teaches that said ionization potential of said at least one electron transporting organic material is larger than an energy necessary for the highest-level electron of said at least one hole transporting material to be taken out to a vacuum infinite far point or an ionization potential of said at least one electron transporting organic material is more than an ionization potential of said at least one hole transporting organic material, by stating, "... the difference in ionization potential I_p between the layer having a hole injecting and transporting function and the layer having a[n] ... electron injecting and transporting function is at least 0.25 eV," (col 10 ln 47-52), in order to have a, "photo-electron function," (col 4 ln 41). Please note that it is well known in the art that the energy necessary for the highest-level electron of a given material to be taken out to a vacuum infinite far point is the very definition of an ionization potential.

14. Therefore it would have been obvious to one of ordinary skill in the art at the time of the invention to use the ionization potentials of Nakaya along with the photodetector of Petritsch/Stossel to have a device with a photo-electron function.

15. Petritsch/Stossel/Nakaya discloses most aspects of the instant invention (see paragraphs 5-14 above). However, Petritsch/Stossel/Nakaya fails to show that said ionization potential of said at least one electron transporting organic material is larger than an ionization potential of said at least one hole transporting organic material by 0.6 eV or more. Nonetheless, the skilled artisan would know too that the difference in ionization potential between the hole and electron transporting materials would impact charge separation effectiveness.

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16. The specific claimed difference in ionization potential, absent any criticality, is only considered to be the “optimum” difference in ionization potential disclosed by Petritsch/Stossel/Nakaya that a person having ordinary skill in the art would have been able to determine using routine experimentation based, among other things, on the desired charge separation, manufacturing costs, etc., (see Goesch, 205 USPQ 215 (CCPA (19080)), and since neither non-obvious nor unexpected results, i.e. results which are different in kind and not in degree from the results of the prior art, will be obtained as long as the difference in ionization potentials is used, as already suggested by Petritsch/Stossel/Nakaya.

17. Since the applicant has not established the criticality (see next paragraph) of the difference in ionization potentials stated and since these differences in ionization potentials are in common use in similar devices in the art, it would have been obvious to one of ordinary skill in the art at the time of the invention to use these values in the device of Petritsch/Stossel/Nakaya.

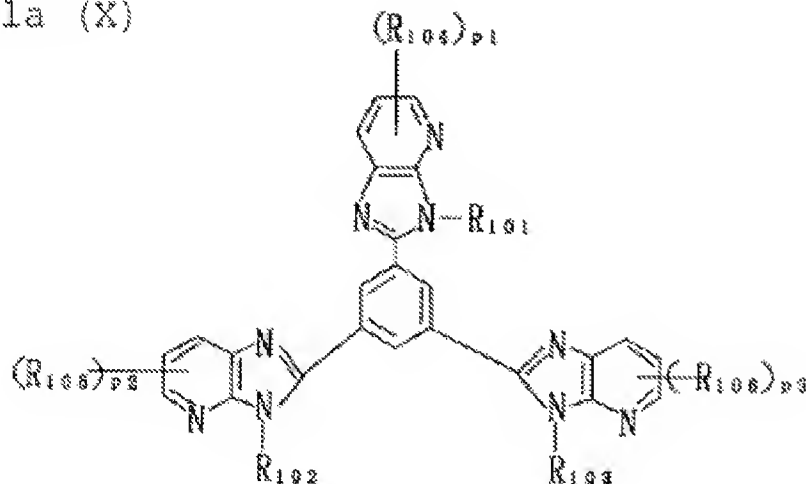
CRITICALITY

18. Please note that the specification contains no disclosure of either the critical nature of the claimed difference in ionization potentials or any unexpected results arising therefrom. Where patentability is said to be based upon particular chosen dimensions or upon another variable recited in a claim, the applicant must show that the chosen dimensions are critical. In re Woodruff, 919 F.2d 1575, 1578, 16 USPQ2d 1934, 1936 (Fed. Cir. 1990).

19. Petritsch/Stossel/Nakaya do not explicitly teach that said at least one electron transporting organic material is a compound represented by formula (X):

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Formula (X)

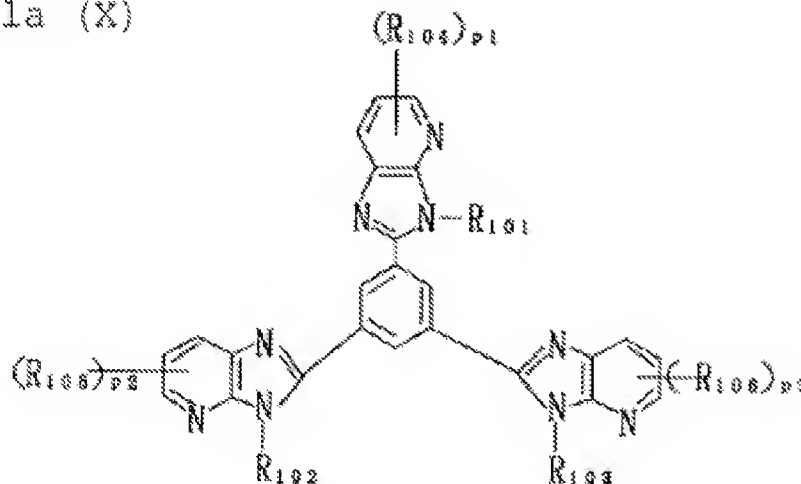


wherein R101, R102 and R103 each represent a hydrogen atom, an aliphatic hydrocarbon group, an aryl group or a hetero ring group, wherein R104, R105, and R106 each independently represents a substituent, wherein the substituent is an alkyl group, an alkenyl group, an alkynyl group, an aryl group, an amino group, an alkoxy group, an arthroxy group, an acyl group, an alkoxycarbonyl group, an aryloxycarbonyl group, an acyloxy group, an acylamino group, an alkoxycarbonylamino group, an aryloxycarbonylamino group, a sulfonylamino group, a sulfamoyl group, a carbamoyl group, an alkylthio group, an arylthio group, a sulfonyl group, a sulfinyl group, a ureido group, a phosphoric acid amido group, a hydroxyl group, a mercapto group, a halogen atom, a cyano group, a sulfo group, a carboxyl group, a nitro group, a hydroxamic acid group, a sulfinio group, a hydrazine group, an imino group, a hetero ring group, or a silyl group, wherein p1, p2 and p3 each independently represents an integer of 0 to 3.

20. It is taught that said at least one electron transporting organic material is a compound represented by formula (X) (General formula (A) at [0021] and [0069]):

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Formula (X)

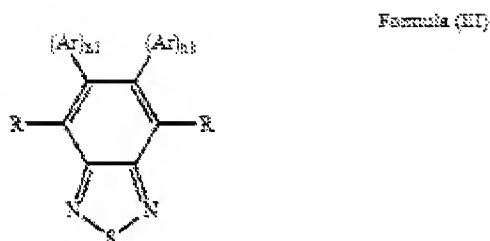


wherein R_{101} , R_{102} and R_{103} each represent a hydrogen atom, an aliphatic hydrocarbon group, an aryl group or a hetero ring group ([0022]), wherein R_{104} , R_{105} , and R_{106} each independently represents a substituent ([0022]), wherein the substituent is an alkyl group, an alkenyl group, an alkynyl group, an aryl group, an amino group, an alkoxy group, an arthroxy group, an acyl group, an alkoxycarbonyl group, an aryloxycarbonyl group, an acyloxy group, an acylamino group, an alkoxycarbonylamino group, an aryloxycarbonylamino group, a sulfonylamino group, a sulfamoyl group, a carbamoyl group, an alkylthio group, an arylthio group, a sulfonyl group, a sulfinyl group, a ureido group, a phosphoric acid amido group, a hydroxyl group, a mercapto group, a halogen atom, a cyano group, a sulfo group, a carboxyl group, a nitro group, a hydroxamic acid group, a sulfinio group, a hydrazine group, an imino group, a hetero ring group, or a silyl group ([0065]), wherein $p1$, $p2$ and $p3$ each independently represents an integer of 0 to 3 ([0022]) to achieve, “the result that blue light can be emitted with high luminance efficiency,” ([0016]).

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21. Therefore it would have been obvious to one of ordinary skill in the art at the time of the invention to use the electron transporting organic material of Ise along with the photodetector of Petritsch/Stossel/Nakaya to achieve blue light emission with high luminance efficiency.

22. With regard to **claim 4**, Stossel teaches the use of 2,1,3-Benzothiadiazole derivatives as Formula (III):



as, “An organic photodetector ... transport material ...” (col 30 ln 11-13).

23. However Stossel does not explicitly disclose the fact that the ionization potential of 2,1,3-Benzothiadiazole derivatives is 6.0 eV or more.

24. Asfandiarov provides evidence that the ionization energies of 2,1,3-benzothiadiazole derivatives are in excess of 6.0 eV, as they range from 7.77 - 8.44 eV (pg 597, Table 2).

25. Thus Stossel does teach that the ionization potential of said at least one electron transporting organic material is 6.0 eV or more.

26. With regard to **claim 14**, Petritsch teaches in Fig 5: at least one transparent electrode (Fig 5 item 4, “are transparent” col 5 ln 32); and at least one electrode (Fig 5 item 12), wherein said at least one electron transporting organic material (Fig 5 item 14) and said at least one hole transporting organic material (Fig 5 item 6) are interposed between said at least one transparent electrode and said at least one electrode (it is clear from Fig 5 that items 14 and 6 are disposed between items 4 and 12).

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27. With regard to **claim 15**, Petritsch/Stossel/Nakaya/Ise teach most of the limitations of this claim as discussed above in paragraphs 5-21. However, Petritsch/Stossel/Nakaya/Ise do not explicitly teach that said at least one electron transporting organic material is deposited in vacuum. Nonetheless, this is considered an intermediate process step that does not affect the structure of the final device. As to the grounds of rejection under section 103, see MPEP § 2113 which discusses the handling of “product-by-process” claims.

28. With regard to **claim 16**, Petritsch/Stossel/Nakaya/Ise teach most of the limitations of this claim as discussed above in paragraphs 5-21. However, Petritsch/Stossel/Nakaya/Ise do not explicitly teach that at least one of said at least one electron transporting organic material and said at least one hole transporting organic material is deposited in vacuum. Nonetheless, this is considered an intermediate process step that does not affect the structure of the final device. As to the grounds of rejection under section 103, see MPEP § 2113 which discusses the handling of “product-by-process” claims.

29. With regard to **claim 17**, Petritsch teaches: An imaging device comprising a photodetector. by stating, “there is provided a method of forming a ... photoconductive device ...” (col 2 ln 24-26).

30. **Claims 18-20** are rejected under 35 U.S.C. 103(a) as being unpatentable over **Petritsch et al (US 6340789)** in view of **Stossel et al (US 7223484)** as evidenced by **Asfandiarov et al (Investigation of Electron Structure of 2,1,3-Benzothiadiazole Derivatives by Means of Negative Ion Mass Spectrometry, Photoelectron Spectroscopy and Absorption Spectroscopy; Rapid Commun. Mass Spectrom. 12, 595–602, 1998)**, **Nakaya et al (US**

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5792557), and **Ise et al. (US 2002/0028329)** as applied to claim 17 above, and further in view of **Iwasaki (US 2003/0209651)**.

31. Petritsch/Stossel/Nakaya/Ise teach most of the limitations of these claims as discussed above in paragraphs 5-21 and 29.

32. Petritsch also teaches in Fig 5: a substrate (Fig 5 item 10); and a first layer comprising a first photodetector (Fig 5 items 6, 28, and 14).

33. Petritsch/Stossel/Nakaya/Ise do not explicitly teach: a second layer comprising a second photodetector; a third layer comprising a third photodetector; the first photodetector comprises a blue light photodetector; the second photodetector comprises a green light photodetector; and the third photodetector comprises a red light photodetector.

34. Iwasaki teaches a second layer comprising a second photodetector (Fig 1 item 101); a third layer comprising a third photodetector (Fig1 item 103); the first photodetector (Fig 1 item 102) comprises a blue light photodetector (clearly visible in Fig 1 as the portion of the light labeled “B” is absorbed by this layer); the second photodetector comprises a green light photodetector (clearly visible in Fig 1 as the portion of the light labeled “G” is absorbed by this layer); and the third photodetector comprises a red light photodetector (clearly visible in Fig 1 as the portion of the light labeled “R” is absorbed by that layer), in order to not have to use a color filter system ([0005]-[0007]).

35. Therefore it would have been obvious to one of ordinary skill in the art at the time of the invention to use the multiple photodetectors of Iwasaki along with the photodetector of Petritsch/Stossel/Nakaya/Ise to not have to use a color filter system.

Response to Arguments

36. Applicant's arguments with respect to claims 1, 4, 12, and 14-20 have been considered but are moot in view of the new ground(s) of rejection.

Conclusion

37. Applicant's amendment necessitated the new ground(s) of rejection presented in this Office action. Accordingly, **THIS ACTION IS MADE FINAL**. See MPEP § 706.07(a). Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

38. A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than SIX MONTHS from the date of this final action.

39. Any inquiry concerning this communication or earlier communications from the examiner should be directed to RAJ GUPTA whose telephone number is (571)270-5707. The examiner can normally be reached on Monday-Thursday 9am-6pm.

40. If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Wael M. Fahmy can be reached on (571)272-1705. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

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41. Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see <http://pair-direct.uspto.gov>. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free). If you would like assistance from a USPTO Customer Service Representative or access to the automated information system, call 800-786-9199 (IN USA OR CANADA) or 571-272-1000.

RAJ GUPTA
Examiner, Art Unit 2814
September 2, 2010

/Marcos D. Pizarro/
Primary Examiner, Art Unit 2814